

# Microstructures of MgB<sub>2</sub>/Fe tapes fabricated by an *in situ* powder-in-tube method using MgH<sub>2</sub> as a precursor powder

S Hata<sup>1,5</sup>, T Yoshidome<sup>1</sup>, H Sosiati<sup>2,3</sup>, Y Tomokiyo<sup>1</sup>, N Kuwano<sup>3</sup>, A Matsumoto<sup>4</sup>, H Kitaguchi<sup>4</sup> and H Kumakura<sup>4</sup>

<sup>1</sup> Department of Applied Sciences for Electronics and Materials, Kyushu University, Kasuga 816-8580, Japan

<sup>2</sup> Research Laboratory for High Voltage Electron Microscopy, Kyushu University, Fukuoka 812-8581, Japan

<sup>3</sup> Art, Science and Technology Centre for Cooperative Research, Kyushu University, Kasuga 816-8580, Japan

<sup>4</sup> National Institute for Materials Science, Tsukuba 305-0047, Japan

E-mail: [hata@asem.kyushu-u.ac.jp](mailto:hata@asem.kyushu-u.ac.jp)

Received 8 November 2005, in final form 30 November 2005

Published 4 January 2006

Online at [stacks.iop.org/SUST/19/161](http://stacks.iop.org/SUST/19/161)

## Abstract

Microstructures of MgB<sub>2</sub>/Fe tapes fabricated by an *in situ* powder-in-tube method using MgH<sub>2</sub> as a precursor powder have been studied by means of x-ray diffraction and analytical transmission electron microscopy combined with a focused ion beam microsampling technique. The overall microstructures in the tapes are characterized as densely crystallized MgB<sub>2</sub> areas with 10–200 nm grain size, uncrystallized areas mainly containing MgO and amorphous phases enriched with B, and a number of holes and cracks. The crystallized MgB<sub>2</sub> areas increase upon doping with SiC nanoparticles. Si and C atoms decomposed from SiC nanoparticles during heat treatment exhibit different spatial distributions: the Si atoms are inhomogeneously distributed, forming silicides such as Mg<sub>2</sub>Si with grain size of 5–20 nm, while the C atoms tend to be uniformly distributed in the MgB<sub>2</sub> matrix. A significant difference in distribution of O atoms between the SiC-doped and non-doped specimens was observed. The processes of formation of these microstructures and their relationships with the critical current density under magnetic fields have been discussed.

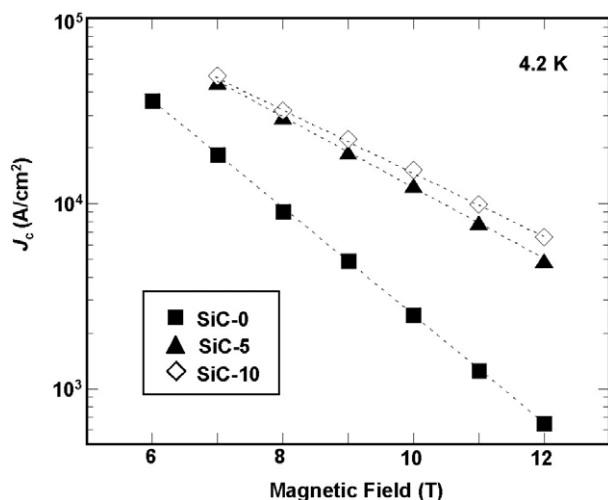
## 1. Introduction

Extensive research efforts as regards superconducting tapes of MgB<sub>2</sub> [1] are being made worldwide, intending to increase the critical current density ( $J_c$ ). The  $J_c$  of *in situ* processed MgB<sub>2</sub> tapes depends sensitively on fabrication parameters, such as heat treatment conditions, starting materials and additional dopants. Since these parameters strongly influence the microstructures as well as the  $J_c$  properties of the *in*

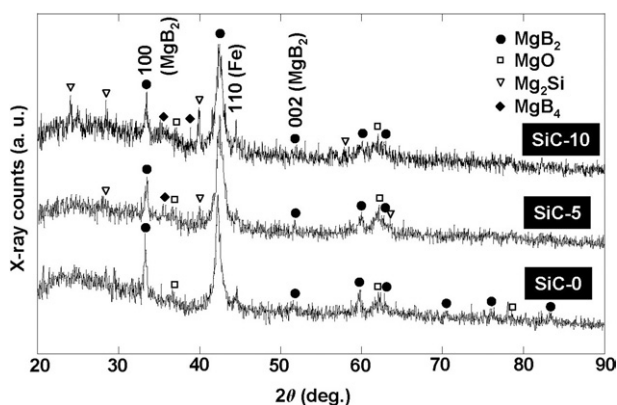
*situ* processed MgB<sub>2</sub> tapes, microstructural analysis helps one to understand the relationships between the fabrication parameters and the  $J_c$  properties.

This paper focuses on MgB<sub>2</sub>/Fe tapes fabricated by an *in situ* powder-in-tube (PIT) method using MgH<sub>2</sub> as a precursor powder [2–6]. It has been reported that use of the MgH<sub>2</sub> powder in the *in situ* PIT process has a number of advantages over use of conventional Mg powder. Firstly,  $J_c$  values of MgB<sub>2</sub> tapes prepared with the MgH<sub>2</sub> powder are approximately double the values for those prepared with the Mg powder. Secondly, heat treatment temperatures can

<sup>5</sup> Author to whom any correspondence should be addressed.



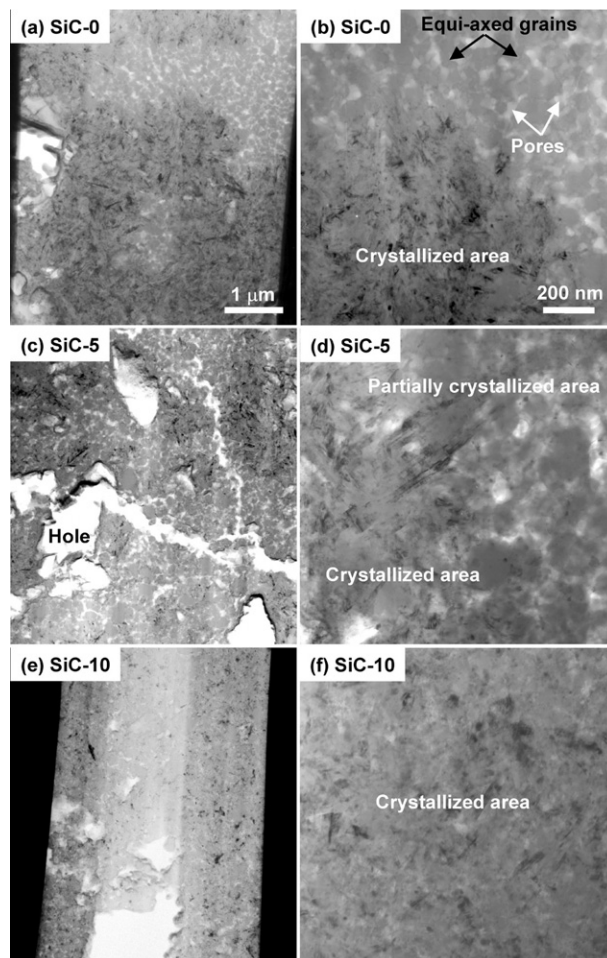
**Figure 1.**  $J_c$  for SiC-0, SiC-5 and SiC-10 as a function of the magnetic field.



**Figure 2.** XRD patterns for SiC-0, SiC-5 and SiC-10.

be reduced without degrading the  $J_c$  values. Doping with impurities such as nanosized SiC [7–11] is also effective for enhancing  $J_c$  under magnetic fields [3–5]. Owing to these advantages, knowledge of microstructures of the  $MgB_2/Fe$  tapes prepared with the  $MgH_2$  powder will be a good guide for developing  $MgB_2$  tapes with excellent  $J_c$  properties. However, no detailed microstructural study of these tapes has been reported.

In the present study, microstructures of the  $MgB_2/Fe$  tapes fabricated by an *in situ* PIT method using  $MgH_2$  as a precursor powder have been investigated in detail by means of x-ray diffraction (XRD) and analytical transmission electron microscopy (ATEM). Since microstructures in  $MgB_2$  tapes and wires are often spatially inhomogeneous [12–15], a focused ion beam (FIB) microsampling technique [16] has been applied for specimen preparation. This technique can be used to prepare TEM specimens from desired areas in the  $MgB_2$  tapes, which is beneficial for correctly characterizing microstructures over the entire area of the tapes. To the best of our knowledge, no other research groups have used this technique effectively for addressing  $MgB_2$  tapes. Nanometre-scale elemental mapping using characteristic x-rays has also been carried out to clarify

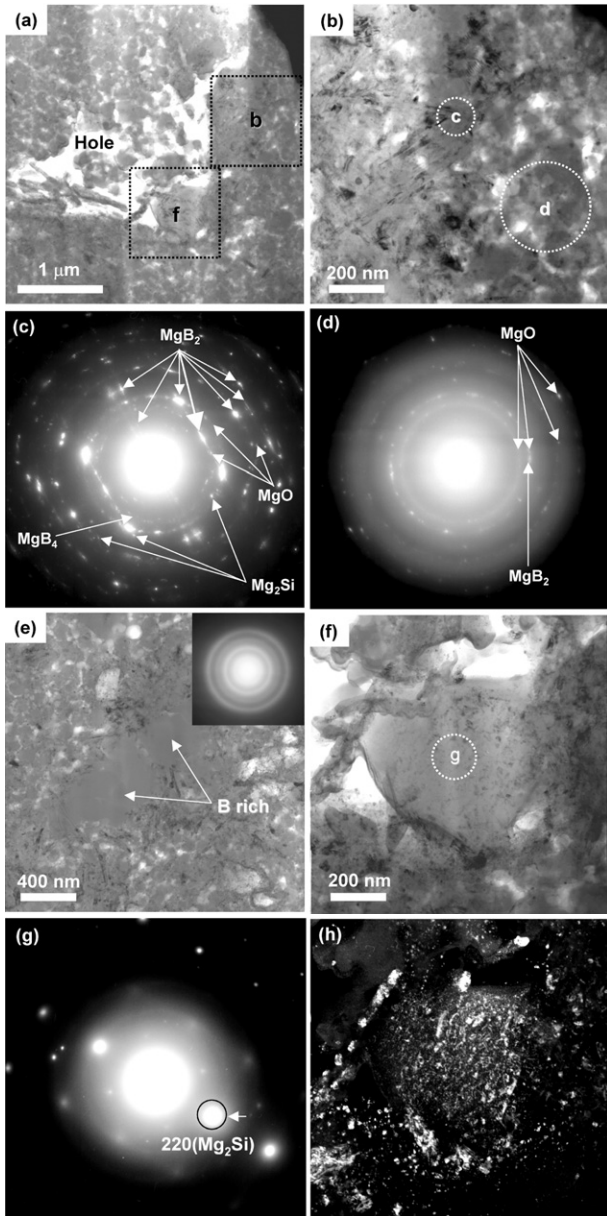


**Figure 3.** Low- and medium-magnification BF TEM images of SiC-0 ((a), (b)), SiC-5 ((c), (d)) and SiC-10 ((e), (f)).

the two-dimensional (2D) distribution of constituent elements in the microstructures. On the basis of the results obtained, the formation of microstructure in the *in situ* PIT process and relationships between the microstructures and  $J_c$  properties will be discussed.

## 2. Experiment

The detailed procedure for preparation of  $MgB_2/Fe$  tapes using  $MgH_2$  as a precursor powder has been reported elsewhere [2–5].  $MgH_2$  powder (325 mesh, 90%) and amorphous B powder (325 mesh, 99.9%) were mixed with the nominal composition ratio,  $Mg:B = 1:2$ . 5 or 10 at.% of SiC powders about 30 nm in size were doped into the mixed powder of  $MgH_2$  and amorphous B. In this paper, the specimens of non-doped  $MgB_2$  tape and SiC-doped  $MgB_2$  tapes with 5 and 10 at.% SiC are designated as SiC-0, SiC-5 and SiC-10, respectively. These mixed powders were packed into pure Fe tubes 5 cm in length, and then cold-rolled into tapes. The final size of the tapes was about 4.5 mm width and 0.5 mm thickness. The tape specimens were heated under an Ar atmosphere up to 873 K for 0.5 h and kept at this temperature for 1 h, and then cooled in a furnace.



**Figure 4.** TEM results for SiC-5. (a) Low-magnification BF image; (b) enlarged image of area b in (a); (c) ED pattern taken from the encircled areas c and d in (b); (d) ED pattern taken from the encircled area d in (b); (e) BF image and an ED pattern of amorphous B-rich areas, (f) BF image of a large grain of Mg<sub>2</sub>Si taken from area f in (a); (g) ED pattern of area g in (f); and (h) DF image obtained using the 220<sub>Mg<sub>2</sub>Si</sub> reflection in (g).

The critical current ( $I_c$ ) of the tape specimens was measured by a four-probe resistive method in a magnetic field at 4.2 K.  $J_c$  was obtained by dividing the measured  $I_c$  by the cross-sectional area of the MgB<sub>2</sub> core. The criterion of  $1 \mu\text{V cm}^{-1}$  was used for  $I_c$  determination. The magnetic field was applied parallel to the tape surface.

Crystalline phases in all specimens were identified from x-ray diffraction (XRD) results. Specimens for the XRD experiment were prepared by mechanically peeling off the MgB<sub>2</sub> core from the tape specimens. An FIB system (HITACHI FB-2000K) equipped with a microsampling unit

**Table 1.** Lattice parameters of MgB<sub>2</sub> determined from the XRD experiment in figure 2.

Specimen	$a_{\text{MgB}_2}$ (nm)	$c_{\text{MgB}_2}$ (nm)
SiC-0	0.3089	0.3533
SiC-5	0.3083	0.3532
SiC-10	0.3079	0.3533

was used for preparing TEM specimens. Microsamples with the size of  $15 \mu\text{m} \times 3 \mu\text{m} \times 15 \mu\text{m}$  were picked up from the middle and edge parts of the MgB<sub>2</sub> core. A TEM (FEI TECNAI-F20) equipped with scanning TEM and energy dispersive x-ray spectroscopy (STEM-EDXS) systems was used for microstructural analysis. In the STEM-EDXS experiments, an electron beam about 1 nm in diameter was positioned on the specimen surface, and an EDX spectrum was acquired for 4 s at each position. Elemental maps were obtained by displaying integrated intensities of characteristic x-rays (B K: 0.18 keV, C K: 0.28 keV, O K: 0.53 keV, Mg K: 1.25 keV, Si K: 1.74 keV) at each of the acquisition points without background subtraction. All the specimens were cleaned with a plasma cleaner (ULVAC PLASMA CUBE C50-MK) before the STEM-EDXS experiments were carried out to prevent hydrocarbon contamination.

### 3. Results

#### 3.1. $J_c$ properties

Figure 1 shows  $J_c$  for SiC-0, SiC-5 and SiC-10 as a function of the magnetic field. In magnetic fields up to 12 T, the  $J_c$  values for SiC-5 and SiC-10 are an order of magnitude higher than that for SiC-0. At the same magnitude of the magnetic field,  $J_c$  for SiC-10 is slightly higher than that for SiC-5. This clearly shows that the addition of nanosized SiC in the MgB<sub>2</sub> tapes has induced a marked enhancement of  $J_c$ , whereas an SiC doping level higher than 5 at.% yielded less pronounced effects on  $J_c$ .

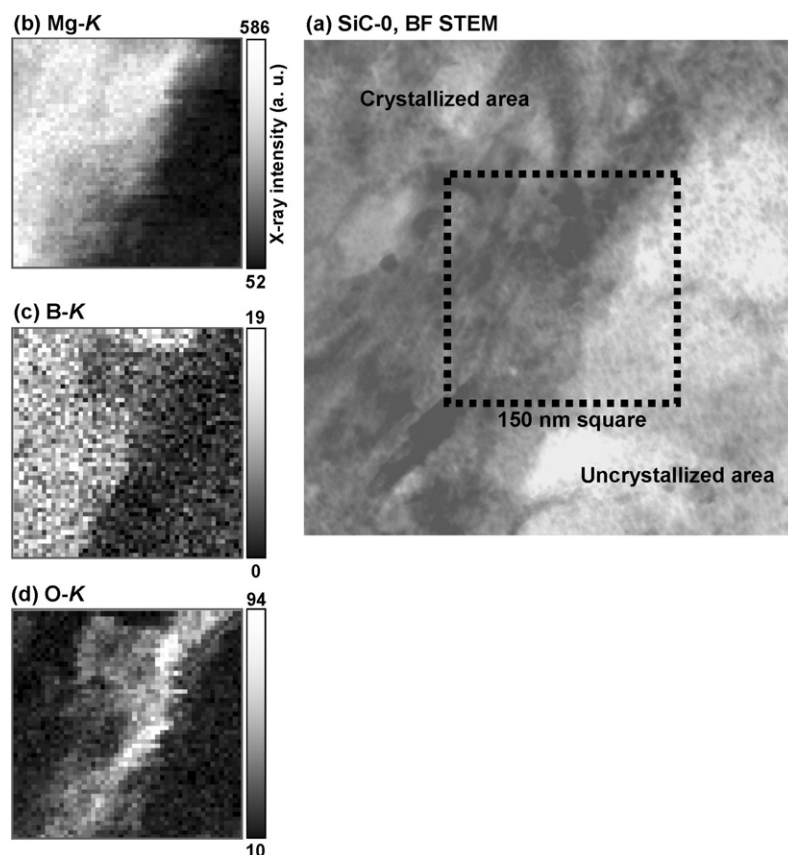
#### 3.2. XRD analysis

The XRD patterns of SiC-0, SiC-5 and SiC-10 in figure 2 show MgB<sub>2</sub> as a main phase and MgO, MgB<sub>4</sub> and Mg<sub>2</sub>Si as impurity phases. The diffraction peaks from Fe seen in all the diffraction patterns are due to mixing of the Fe sheath material in the specimen preparation. No diffraction peak from SiC was identified, which suggests that the nanosized SiC was decomposed during the heat treatment. These XRD patterns also indicate that the Bragg angles of the reflections from MgB<sub>2</sub> slightly change with the SiC doping. The lattice parameters,  $a$  and  $c$ , of the hexagonal MgB<sub>2</sub> crystal were calculated from these Bragg angles, and these are shown in table 1. It is noted that the lattice parameter,  $a$ , decreases with increasing SiC content. This suggests that some of the B atoms in the MgB<sub>2</sub> crystal have been replaced with C atoms decomposed from the SiC nanoparticles [7, 9–11, 17, 18].

#### 3.3. TEM analysis

Figures 3(a), (c) and (e) show bright-field (BF) TEM images of microsamples of SiC-0, SiC-5 and SiC-10, respectively.



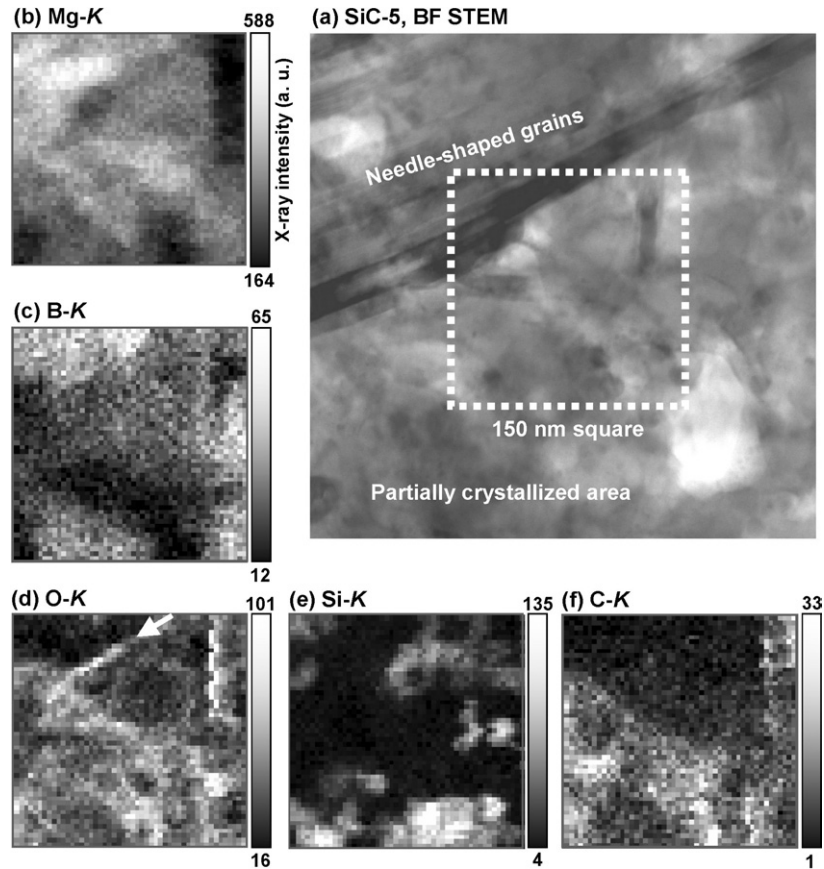


**Figure 5.** STEM-EDXS elemental mapping for SiC-0. (a) BF STEM image; (b) Mg K map; (c) B K map; and (d) O K map.

These microsamples were prepared from the middle of the  $\text{MgB}_2$  core. All the specimens contain a number of holes and cracks, as reported by other researchers [12, 13]. Scanning electron microscopy observations confirmed that these holes and cracks were originally formed in the *in situ* PIT process. A magnified image of SiC-0 in figure 3(b) shows two different microstructures. One microstructure in the lower part of the image exhibits dark contrast due to diffraction in the crystals. Another microstructure in the upper part is composed of equi-axed grains with grey contrast and small pores with bright contrast, and similar microstructures have been observed by Li *et al* [19]. As described in detail later, the former microstructure corresponds to crystallized  $\text{MgB}_2$  areas, and the latter corresponds to uncrystallized areas that mainly contain MgO and amorphous phases with B-rich compositions. The dark diffraction contrast in the crystallized areas shows that some of the  $\text{MgB}_2$  crystals have needle-like shapes. This morphology of the  $\text{MgB}_2$  crystals is a feature of the  $\text{MgB}_2$  tapes prepared with the  $\text{MgH}_2$  powder, as reported previously [3]. Figure 3(d) shows a magnified image of SiC-5. Dark diffraction contrast that suggests the crystallization of  $\text{MgB}_2$  is observed in some of the uncrystallized areas. In the case of SiC-10 in figure 3(f), the crystallized areas have significantly increased, resulting in a homogeneous microstructure in the  $\text{MgB}_2$  core. The grain size of  $\text{MgB}_2$  is about 10–200 nm and not clearly dependent upon the SiC content.

Figure 4 shows the result of detailed TEM observation for SiC-5, as a representative of the three specimens: SiC-0, SiC-

5 and SiC-10. A low-magnification BF image in figure 4(a) shows the inhomogeneous microstructure at the sub- $\mu\text{m}$  scale as described above. A magnified image, figure 4(b), of the area b in figure 4(a) shows a mixture of crystallized and uncrystallized areas. An electron diffraction (ED) pattern, figure 4(c), taken from the encircled area c in figure 4(b) implies that the crystallized area contains  $\text{MgB}_2$  as the main phase and MgO,  $\text{Mg}_2\text{Si}$  and  $\text{MgB}_4$  as impurity phases. This agrees with the XRD result shown in figure 2. Since diffraction intensities from these crystalline phases appear very close to each other in the ED pattern, dark-field (DF) imaging of each of these phases is difficult. However, it is noted that some of the diffraction spots from  $\text{MgB}_2$ , for example,  $hkl = 100_{\text{MgB}_2}$  indicated by the extra large arrow in figure 4(c), exhibit streaks. This indicates that the needle-shaped grains observed in the crystallized areas are of  $\text{MgB}_2$ . An ED pattern, figure 4(d), from the encircled area d in figure 4(b) shows the existence of MgO and amorphous phases with a small amount of  $\text{MgB}_2$  in the uncrystallized area. EDXS experiments revealed that these uncrystallized areas have higher B and lower Mg concentrations than those in the crystallized areas. Therefore, it turns out that the amorphous phase is enriched with B. As a matter of fact, large grains of the amorphous B were recognized in the uncrystallized areas, as shown in figure 4(e). A magnified image, figure 4(f) of the area f in figure 4(a) shows a large grain of  $\text{Mg}_2\text{Si}$  about 500 nm in size, which was clearly identified from the ED pattern and the DF image in figures 4(g)



**Figure 6.** STEM-EDXS elemental mapping for SiC-5. (a) BF STEM image; (b) Mg K map; (c) B K map; (d) O K map; (e) Si K map; and (f) C K map.

and (h), respectively. No diffraction intensities relating to carbon (e.g. graphite) or carbide phases were recognized.

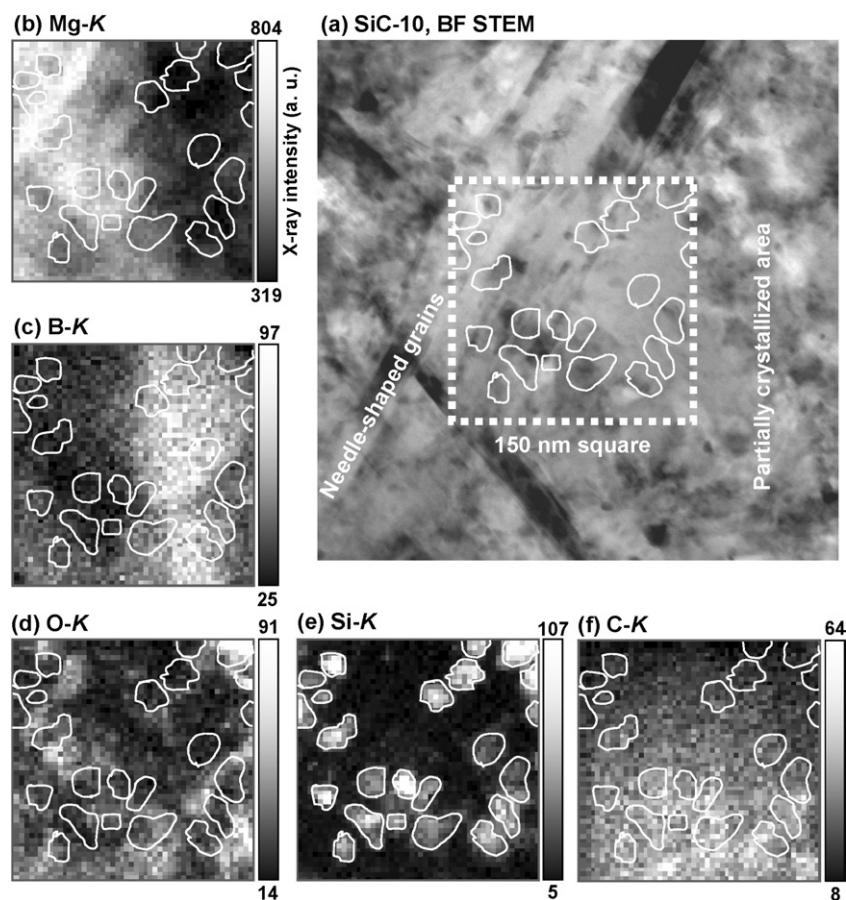
### 3.4. STEM-EDXS analysis

Figure 5 shows a result from the STEM-EDXS elemental mapping for Si-0. The broken open square in the BF STEM image, figure 5(a), indicates a mapping area 150 nm, square, in size, which contains a boundary between crystallized and uncrystallized areas. Since the uncrystallized area has very low density, elemental mapping data do not give straightforward information on the compositions in the two different areas. For example, very low concentrations of Mg (figure 5(b)) and B (figure 5(c)) atoms in the uncrystallized area are mainly due to the very low density in this area. Nevertheless, it is clearly seen that the 2D distribution of the constituent elements is well correlated with the microstructures. The O map (figure 5(d)) clearly shows enrichment of O atoms along the boundary between the crystallized and uncrystallized areas. It was found from TEM observations that MgO was formed in this O-rich area.

Figure 6 shows an elemental mapping result for SiC-5. The mapping area contains a boundary between a crystallized area with needle-shaped grains of MgB<sub>2</sub> and a partially crystallized area without needle-shaped grains. The enrichment of O atoms at the boundary between the crystallized and uncrystallized areas is recognized, as indicated by

the arrow in figure 6(d). However, this enrichment of O atoms is less remarkable than that in the non-doped specimen. Si (figure 6(e)) and C (figure 6(f)) atoms show different distributions. This demonstrates that most of the SiC nanoparticles decomposed during the *in situ* PIT process [3–11]. The decomposed Si atoms are well localized in small areas 5–50 nm in size. This local enrichment of Si atoms is the most significant among those for all the impurity atoms, O, Si and C. The distribution of C atoms (figure 6(f)) is similar to that of O atoms (figure 6(d)) and opposite to that of B atoms (figure 6(c)). This is interpreted as indicating the formation of oxides such as Mg(B, O) [20, 21] and BO<sub>x</sub> [22] or substitution of O and C atoms for B atoms in the MgB<sub>2</sub> matrix [7, 9–11, 17, 18, 23, 24].

Figure 7 shows an elemental mapping result for SiC-10. The mapping area again contains a crystallized area with needle-shaped grains and a partially crystallized area. The local Si-rich areas with the size of 5–20 nm increase in number with the SiC content, as depicted with white lines in figure 7(e). These white lines showing the Si-rich areas are also drawn in the other figures for the phase identification described below. It should be noted that O atoms in figure 7(d) tend to be enriched close to the Si-rich areas. On the other hand, C atoms in figure 7(f) tend to disperse uniformly irrespective of the distribution of the other elements. This may mean the substitution of C atoms for B atoms in the MgB<sub>2</sub> crystals, as expected from the XRD experiment. In this elemental mapping



**Figure 7.** STEM-EDXS elemental mapping for SiC-10. (a) BF STEM image; (b) Mg K map; (c) B K map; (d) O K map; (e) Si K map; and (f) C K map.

for SiC-10, one can identify locations of some impurity phases by comparing the elemental maps with each other. For example, most of the Si-rich areas show lower Mg and lower O concentrations. These areas can be identified as  $\text{Mg}_2\text{Si}$  since the density of Mg atoms is lower in the  $\text{Mg}_2\text{Si}$  crystal (cubic  $\text{CaF}_2$  type,  $a = 0.6351$  nm) than in the  $\text{MgB}_2$  crystal (hexagonal  $\text{AlB}_2$  type,  $a = 0.3086$  nm,  $c = 0.3524$  nm). The other Si-rich areas where the above interpretation for  $\text{Mg}_2\text{Si}$  is not suitable may be nanosized silicides such as  $\text{SiO}_2$  and  $\text{Si}_x\text{B}_y\text{O}_z$  [10].

Figure 8 shows another elemental mapping result for SiC-10. The mapping area contains part of a large grain of  $\text{Mg}_2\text{Si}$ . Mg (figure 8(a)) and O (figure 8(d)) atoms are enriched at the peripheries of the  $\text{Mg}_2\text{Si}$  grain, suggesting the formation of MgO.

## 4. Discussion

### 4.1. Sample preparation by the FIB microsampling technique

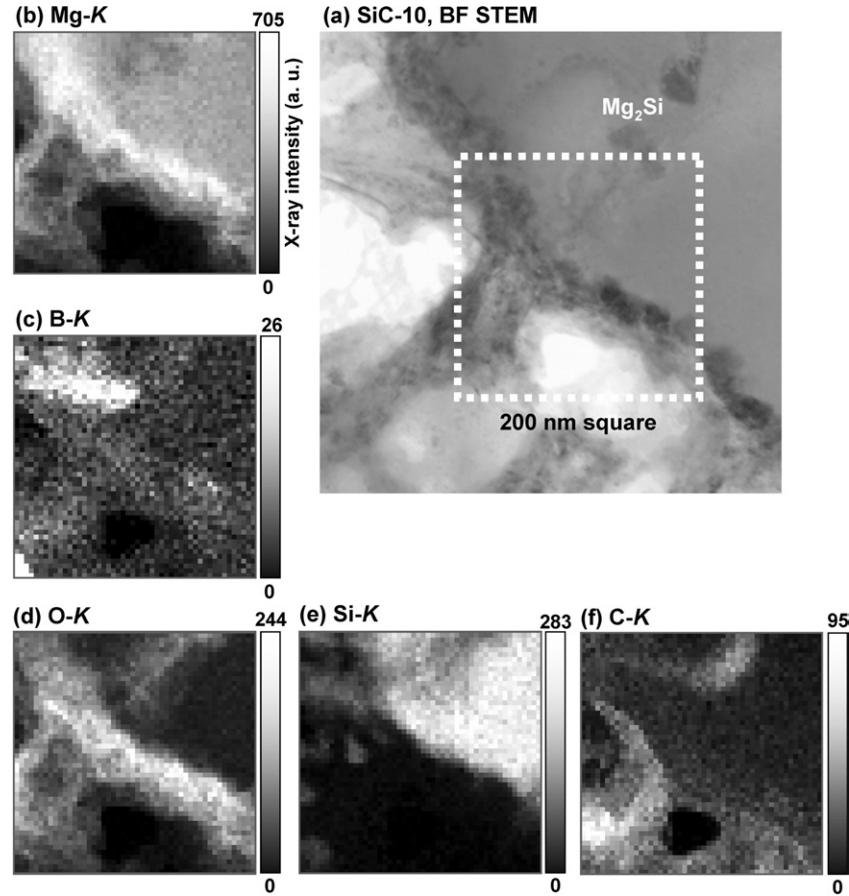
It is emphasized that the FIB microsampling technique is effective for preparing TEM specimens of the present  $\text{MgB}_2/\text{Fe}$  tapes with inhomogeneous microstructures. The TEM observation combined with this technique revealed the existence of holes and cracks and the distribution of crystallized and uncrystallized areas at the sub- $\mu\text{m}$  scale.

As a result, influences of the nanosized SiC doping on these microstructures have been clearly observed. For a conventional TEM specimen preparation method using Ar ion milling, this kind of microstructure analysis is difficult since the uncrystallized areas with low density are preferentially milled and hardly any of the original inhomogeneous microstructure will remain.

### 4.2. Microstructure formation in an *in situ* PIT process

All the  $\text{MgB}_2/\text{Fe}$  tapes prepared with the  $\text{MgH}_2$  powder show characteristic microstructures composed of crystallized  $\text{MgB}_2$  areas and uncrystallized areas. The crystallized areas have compact granular structures of the  $\text{MgB}_2$  crystals, which may be advantageous for increasing  $J_c$  [2–6]. The uncrystallized areas where the reaction for forming  $\text{MgB}_2$  has not been completed are porous and mainly contain MgO and amorphous B-rich phases. In the case of the non-doped  $\text{MgB}_2/\text{Fe}$  tape, significant enrichment of oxygen due to the formation of MgO was observed along the boundary between the crystallized and uncrystallized areas. Since the present  $\text{MgB}_2/\text{Fe}$  tape specimens have been prepared using  $\text{MgH}_2$  powder as a starting material, MgO may be produced in the air after the *in situ* PIT process. In other words, Mg-rich regions that would prevent crystallization of  $\text{MgB}_2$  have been formed at the growth front of the crystallized areas in the non-doped  $\text{MgB}_2/\text{Fe}$  tape.





**Figure 8.** STEM-EDXS elemental mapping at a large grain of Mg<sub>2</sub>Si in SiC-10. (a) BF STEM image; (b) Mg K map; (c) B K map; (d) O K map; (e) Si K map; and (f) C K map.

The crystallized area of MgB<sub>2</sub> in the tapes significantly increases with nanosized SiC doping. The boundaries between the crystallized and uncrystallized areas become ambiguous, and the preferential formation of MgO along the boundaries is suppressed. O atoms in the SiC-doped tapes tend to be uniformly distributed or enriched near Si-rich areas that are mainly of Mg<sub>2</sub>Si. That is, the doping of SiC nanoparticles brings about changes in the distribution of O atoms in the MgB<sub>2</sub> tapes as well as the formation of nanosized silicides. A probable mechanism for the microstructure formation in the SiC-doped tapes would be as follows: in the early stages of the heat treatment that is described above, the SiC nanoparticles decompose into Si and C atoms; the decomposed Si atoms react with Mg atoms to form fine particles of Mg<sub>2</sub>Si before starting the crystallization of MgB<sub>2</sub> [3, 5, 6]; the reaction for forming Mg<sub>2</sub>Si consumes Mg atoms so the preferential enrichment of Mg atoms at the growth front of MgB<sub>2</sub> crystals can be suppressed. There is another possibility: that Mg-rich areas are formed around the Mg<sub>2</sub>Si phase, as shown in figures 7 and 8, since the density of Mg atoms in the Mg<sub>2</sub>Si phase is lower than that in the MgB<sub>2</sub> phase. Since these Mg-rich areas are small in size and dispersed with the nanosized Mg<sub>2</sub>Si, their influences on the crystallization of MgB<sub>2</sub> will not be serious. As a result, the crystallization of MgB<sub>2</sub> proceeds well in the SiC-doped MgB<sub>2</sub> tapes. It is also expected that the fine Mg-rich areas in the SiC-doped tapes may be oxidized

after the *in situ* PIT process and act as effective pinning centres. The decomposed C atoms, on the other hand, partly substitute for B atoms in the MgB<sub>2</sub> crystals, as supported by the XRD and STEM-EDXS data. This substitution behaviour enhances the upper critical field ( $B_{c2}$ ) and contributes to the  $J_c$  increase under high magnetic fields [4].

#### 4.3. Relationships between microstructures and $J_c$ properties

As shown in figure 1, the present MgB<sub>2</sub>/Fe tapes exhibit marked improvement of  $J_c$  under magnetic fields caused by the nanosized SiC doping. It is apparent that the improvement of  $J_c$  characteristics is correlated with the following microstructural changes with the SiC doping: (i) increase of the densely crystallized area of MgB<sub>2</sub>; (ii) formation of nanosized silicides such as Mg<sub>2</sub>Si; (iii) substantial dispersion of nanosized oxides such as MgO; and (iv) C substitution for B atoms in the MgB<sub>2</sub> crystals. The microstructural change (i) improves the flow of supercurrent, and (ii) and (iii) are well known to yield effective pinning centres for MgB<sub>2</sub> materials. The effect of (iv), from the SiC doping, has also been reported previously [4, 7, 10, 11]. These microstructural changes are more pronounced for SiC-10 than for SiC-5. However, the differences in  $J_c$  value between SiC-5 and SiC-10 are not so large, as shown in figure 1. Therefore, the excess doping of SiC nanoparticles is not the best way to effectively improve the  $J_c$  properties. A better way

to further improve  $J_c$  may be reducing the grain sizes of  $\text{MgB}_2$  and impurity phases in order to increase the density of pinning centres such as grain boundaries and nanosized impurities, as shown in the microstructural study of  $\text{MgB}_2$  films [25–30].

## 5. Summary

Microstructures of  $\text{MgB}_2/\text{Fe}$  tapes fabricated by an *in situ* PIT method using  $\text{MgH}_2$  as a precursor powder have been investigated in detail. The results obtained are summarized as follows.

5 or 10 at.% doping of nanosized SiC in the  $\text{MgB}_2/\text{Fe}$  tape results in significant enhancement of  $J_c$ : for example, from  $6 \times 10^2 \text{ A cm}^{-2}$  at 4.2 K and 12 T for the non-doped tape to  $5 \times 10^3 \text{ A cm}^{-2}$  for the 5 at.% SiC-doped one.

By using the FIB microsampling technique for TEM specimen preparation, the microstructures of the  $\text{MgB}_2/\text{Fe}$  tapes are clearly characterized as densely crystallized  $\text{MgB}_2$  areas with 10–200 nm grain size, uncrystallized areas mainly containing MgO and amorphous B-rich phases, and a number of holes and cracks. The nanosized SiC doping increases the crystallized areas and leads to the formation of homogeneous granular structures of  $\text{MgB}_2$ .

STEM-EDXS elemental mapping revealed the enrichment of O atoms due to the formation of MgO at the boundaries between the crystallized and uncrystallized areas. This enrichment of O atoms at the boundary areas is suppressed by SiC doping. O-rich areas in the SiC-doped  $\text{MgB}_2/\text{Fe}$  tapes are small in size and tend to be dispersed near Si-rich areas.

The increase of the densely crystallized area of  $\text{MgB}_2$  improves the flow of supercurrent, and the formation of silicides, mainly  $\text{Mg}_2\text{Si}$ , with grain sizes of about 5–20 nm and the substantial dispersion of nanosized oxides such as MgO will raise the density of effective pinning centres. C substitution for B atoms in the  $\text{MgB}_2$  crystals is supported by XRD and STEM-EDXS data for the SiC-doped specimens. These microstructural changes with SiC doping result in a marked improvement of  $J_c$  under magnetic fields.

## Acknowledgments

This work was supported in part by Nanotechnology Support Project of the Ministry of Education, Culture, Sports, Science and Technology (MEXT), Japan and Research Promotion Bureau, MEXT, Japan under contracts No 16-556.

## References

- [1] Nagamatsu J, Nakagawa N, Muranaka T, Zenitani Y and Akimitsu J 2001 *Nature* **410** 63–4
- [2] Fujii H, Togano K and Kumakura H 2002 *Supercond. Sci. Technol.* **15** 1571–6
- [3] Matsumoto A, Kumakura H, Kitaguchi H and Hatakeyama H 2003 *Supercond. Sci. Technol.* **16** 926–30
- [4] Kumakura H, Kitaguchi H, Matsumoto A and Hatakeyama H 2004 *Appl. Phys. Lett.* **84** 3669–71
- [5] Matsumoto A, Kumakura H, Kitaguchi H and Hatakeyama H 2004 *Supercond. Sci. Technol.* **17** S319–23
- [6] Jiang C H, Hatakeyama H and Kumakura H 2005 *Supercond. Sci. Technol.* **18** L17–22
- [7] Dou S X, Soltanian S, Horvat J, Wang X L, Zhou S H, Ionescu M, Liu H K, Munroe P and Tomsic M 2002 *Appl. Phys. Lett.* **81** 3419–21
- [8] Wang X L, Yao Q W, Horvat J, Qin M J and Dou S X 2004 *Supercond. Sci. Technol.* **17** L21–4
- [9] Wang X L, Soltanian S, James M, Qin M J, Horvat J, Yao Q W, Liu H K and Dou S X 2004 *Physica C* **408–410** 63–7
- [10] Dou S X, Braccini V, Soltanian S, Klie R, Zhu Y, Li S, Wang X L and Larbalestier D 2004 *J. Appl. Phys.* **96** 7549–55
- [11] Soltanian S, Wang X L, Hovart J, Dou S X, Sumption M D, Bhatia M, Collings E W, Monroe P and Tomsic M 2005 *Supercond. Sci. Technol.* **18** 658–66
- [12] Serquis A *et al* 2002 *J. Appl. Phys.* **92** 351–6
- [13] Serquis A, Civale L, Hammon D L, Liao X Z, Coulter J Y, Zhu Y T, Peterson D E and Mueller F M 2003 *J. Appl. Phys.* **94** 4024–31
- [14] Eyidi D, Eibl O, Wenzel T, Nickel K G, Giovannini M and Saccone A 2003 *Micron* **34** 85–96
- [15] Eyidi D, Eibl O, Wenzel T, Nickel K G, Schlachter S I and Goldacker W 2003 *Supercond. Sci. Technol.* **16** 778–88
- [16] Ishitani T, Umemura K, Ohnishi T, Yaguchi T and Kamino T 2004 *J. Electron Microscop.* **53** 443–9
- [17] Dou S X, Yeoh W K, Horvat J and Ionescu M 2003 *Appl. Phys. Lett.* **83** 4996–8
- [18] Cava R J, Zandbergen H W and Inumaru K 2003 *Physica C* **385** 8–15
- [19] Li S, Prabhakar O, Tan T T, Sun C Q, Wang X L, Soltanian S, Horvat J and Dou S X 2002 *Appl. Phys. Lett.* **81** 874–6
- [20] Liao X Z, Serquis A C, Zhu Y T, Huang J Y, Peterson D E, Mueller F M and Xu H F 2002 *Appl. Phys. Lett.* **80** 4398–400
- [21] Liao X Z, Serquis A, Zhu Y T, Huang J Y, Civale L, Peterson D E and Mueller F M 2003 *J. Appl. Phys.* **93** 6208–15
- [22] Klie R F, Idrobo J C, Browning N D, Regan K A, Rogado N S and Cava R J 2001 *Appl. Phys. Lett.* **79** 1837–9
- [23] Klie R F, Idrobo J C, Browning N D, Serquis A, Zhu Y T, Liao X Z and Mueller F M 2002 *Appl. Phys. Lett.* **80** 3970–2
- [24] Klie R F and Zhu Y 2005 *Micron* **36** 219–31
- [25] Sosiati H, Hata S, Kuwano N, Tomokiyo Y, Matsumoto A, Fukutomi M, Kitaguchi H, Komori K and Kumakura H 2004 *Physica C* **412–414** 1376–82
- [26] Hata S, Sosiati H, Tomokiyo Y, Kuwano N, Matsumoto A, Fukutomi M, Kitaguchi H, Komori K and Kumakura H 2004 *J. Japan Inst. Metals* **68** 648–55
- [27] Hata S, Sosiati H, Kuwano N, Tomokiyo Y, Matsumoto A, Fukutomi M, Kitaguchi H, Komori K and Kumakura H 2005 *IEEE Trans. Appl. Supercond.* **15** 3238–41
- [28] Kitaguchi H, Matsumoto A, Kumakura H, Doi T, Yamamoto H, Saitoh K, Sosiati H and Hata S 2004 *Appl. Phys. Lett.* **85** 2842–4
- [29] Kitaguchi H, Doi T, Kobayashi Y, Matsumoto A, Sosiati H, Hata S, Fukutomi M and Kumakura H 2005 *IEEE Trans. Appl. Supercond.* **15** 3313–6
- [30] Sosiati H *et al* 2005 *Supercond. Sci. Technol.* **18** 1275–9