

LAMPIRAN

Lampiran 1. Perhitungan bahan**a. Pembuatan HCl 4N**

HCl pekat dipasaran 37 % v/v

$$37 \% \text{ v/v} = 37 \text{ ml}/100 \text{ ml} = 0,037 \text{ L/ml}$$

b. Perhitungan gram HCl

Bobot larutan HCl = 0,037 L/ml

Berat jenis HCl = 1,190 g/L

$$\begin{aligned} \text{Bobot larutan HCl} \times \text{berat jenis HCl} &= 0,037 \text{ L} \times 1,190 \text{ g/L} \\ &= 440,03 \text{ g} \end{aligned}$$

c. Perhitungan N HCl

$N = \text{g/BM} \times \text{Volume Larutan}$

$$= 440,03 \text{ g}/36,5 \times 0,1 \text{ L}$$

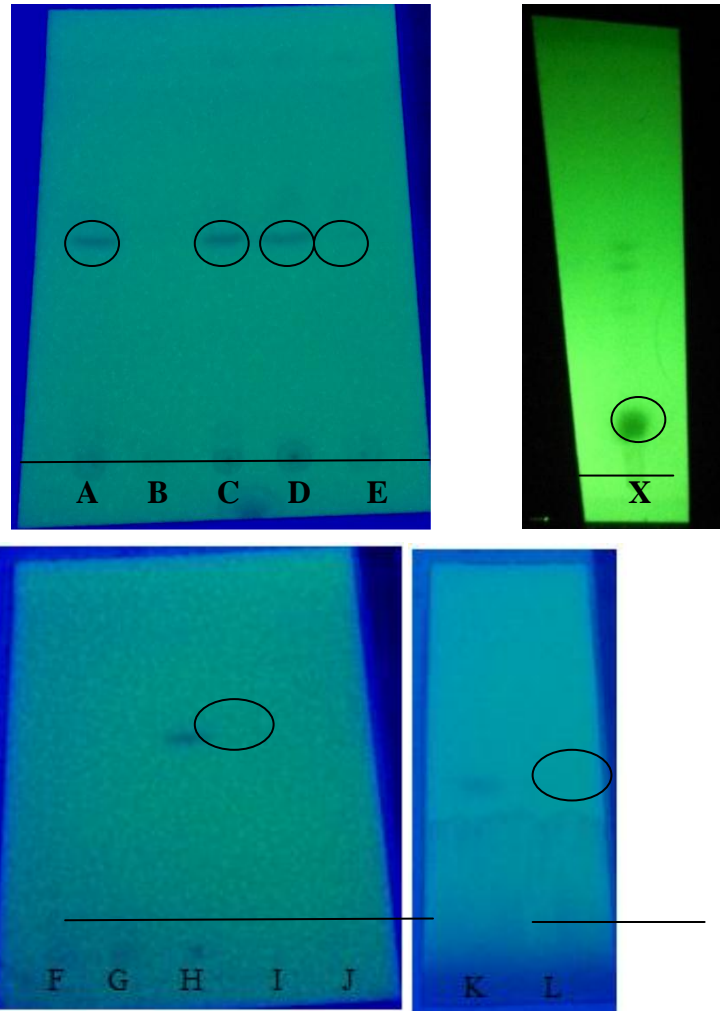
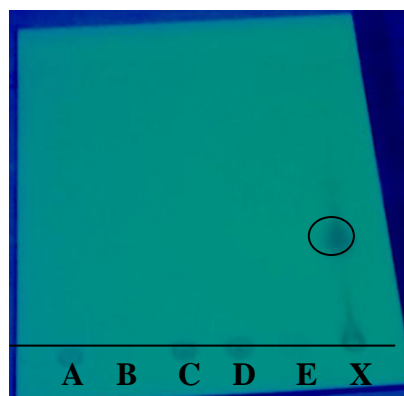
$$= 12,06 \text{ N}$$

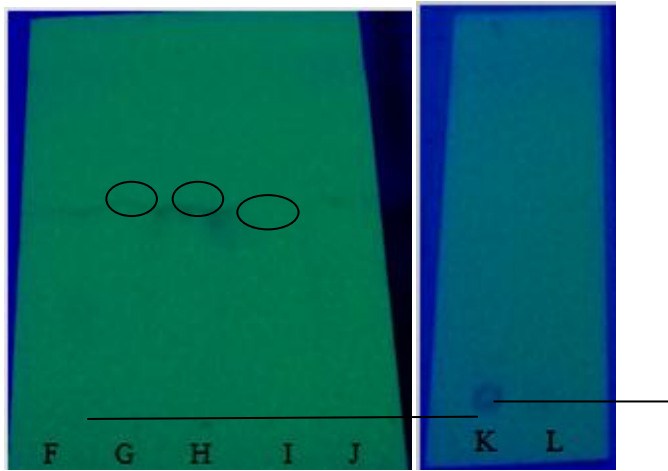
d. Pengenceran HCl

$$N_1 \times V_1 = N_2 \times V_2$$

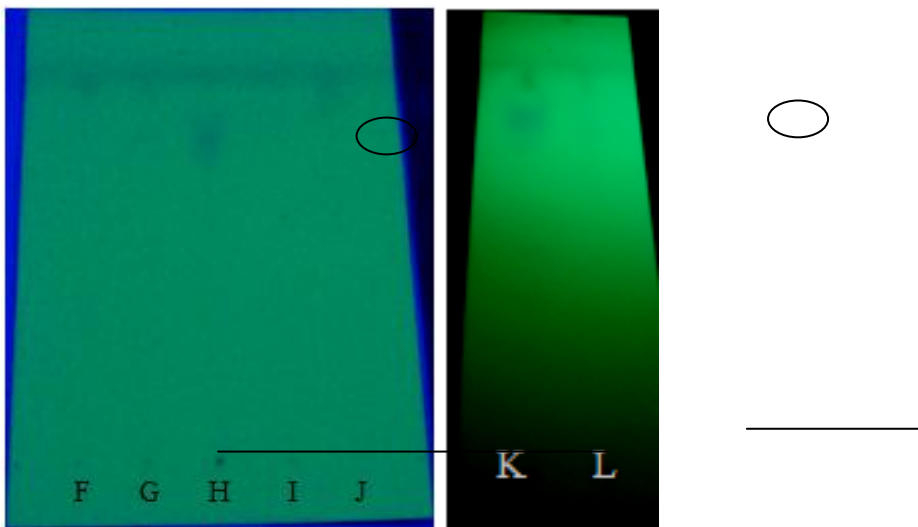
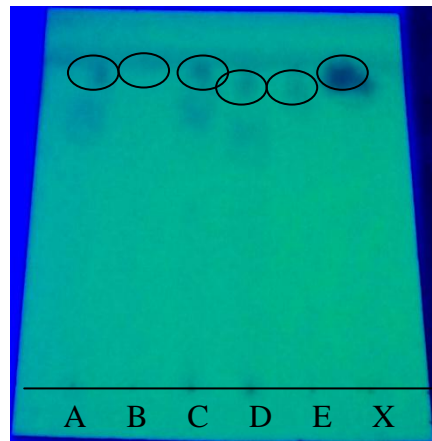
$$12,06 \times V_1 = 4 \times 1000$$

$$V_1 = 331,6/\text{L} = 16,5/50 \text{ mL}$$

Lampiran 2. Uji kualitatif Hidrokinon menggunakan KLT**a. Fase gerak toluen : asam asetat glasial (8:2)****b. Fase gerak n-heksan : aseton (3:2)**



c. Fase gerak kloroform : metanol (5:5)

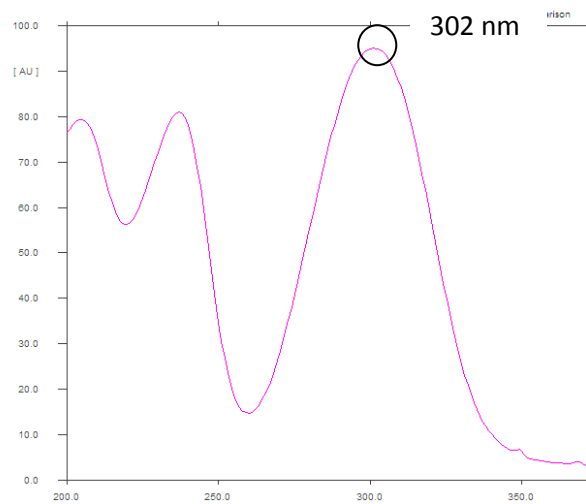


Keterangan :

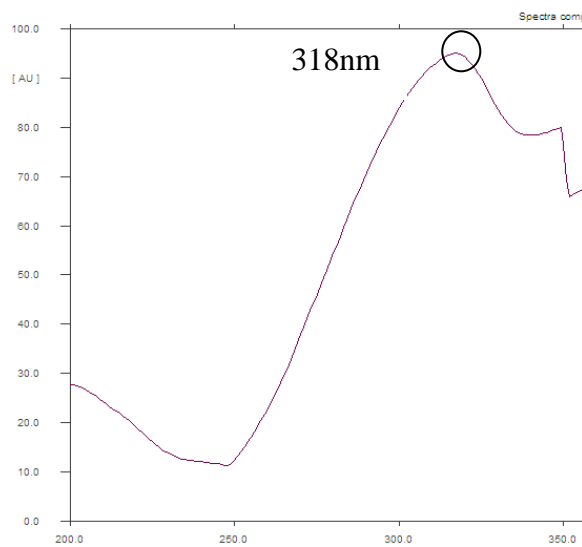
- A : Sampel A
- B : Sampel B
- C : Sampel C
- D : Sampel D
- E : Sampel E
- F : Sampel F
- G : Sampel G
- H : Sampel H
- I : Sampel I
- J : Sampel J
- K : Sampel K
- L : Sampel L
- X : Baku Hidrokinon

Lampiran 3. Panjang Gelombang sampel menggunakan Densitometri

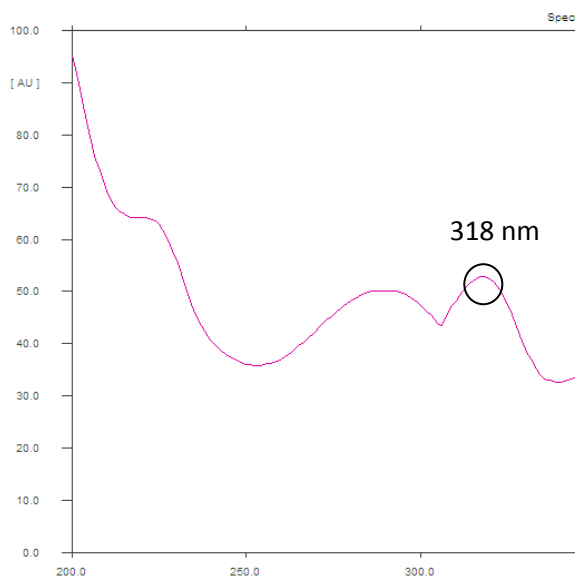
a. Sampel A



Sampel A pada fase gerak toluen:asam asetat glasial

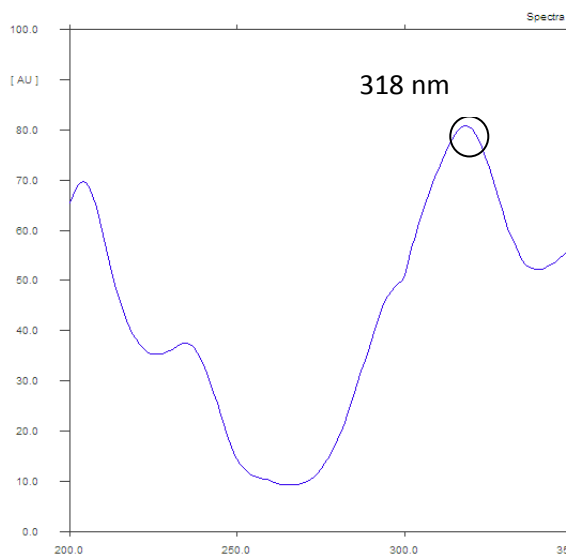


Sampel A pada fase gerak n-heksan:aseton

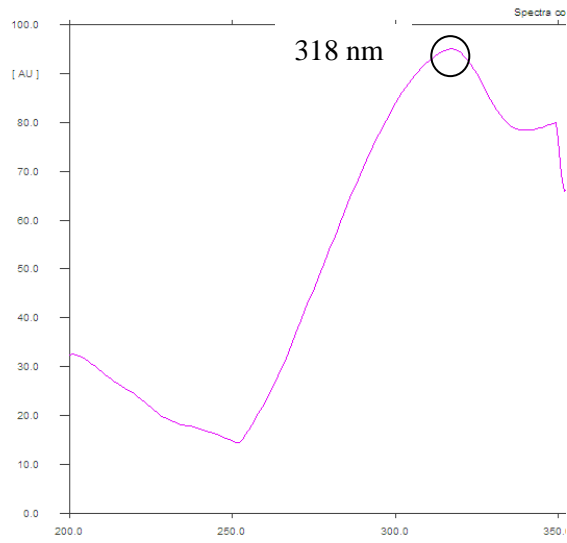


Sampel A pada fase gerak klorofom : metanol

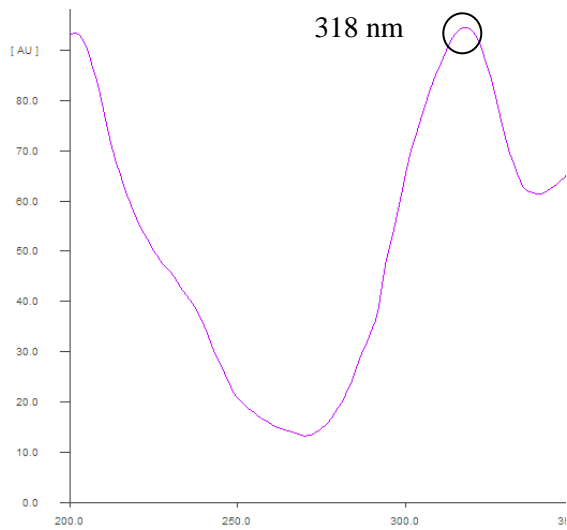
b. Sampel B



Sampel B pada fase gerak toluen:asam asetat glasial

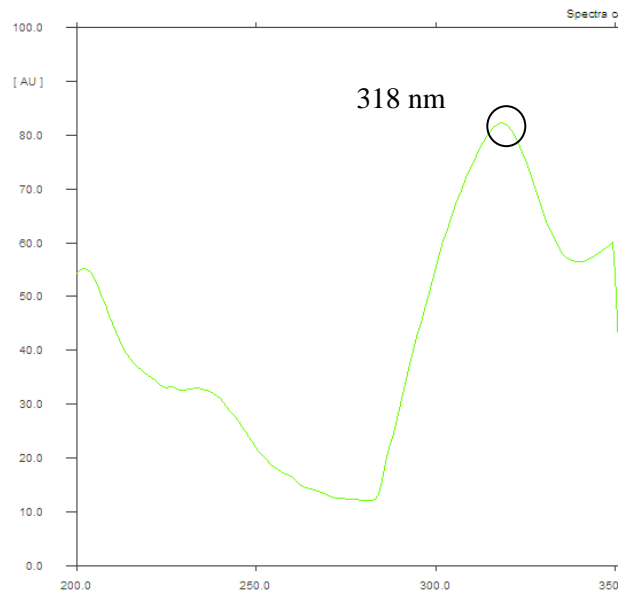


Sampel B pada fase gerak n-heksan:aseton

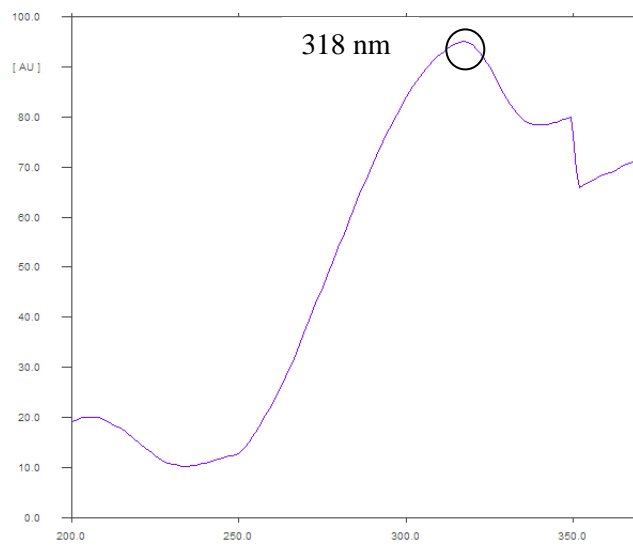


Sampel B pada fase gerak klorofom : metanol

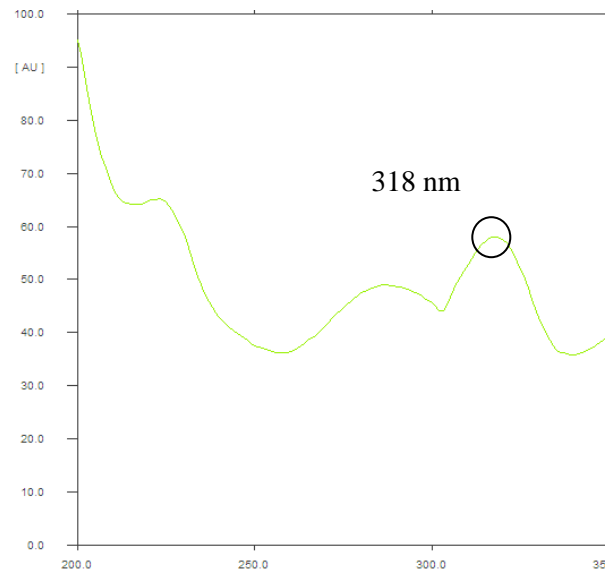
c. Sampel C



Sampel C pada fase gerak toluen:asam asetat glasial

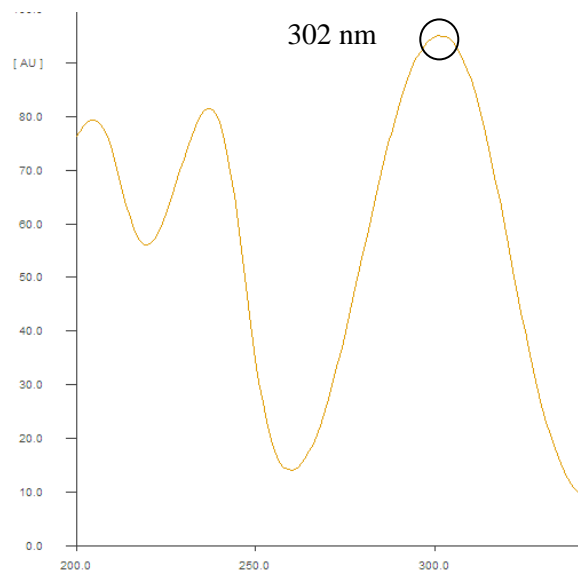


Sampel C pada fase gerak n-heksan:aseton

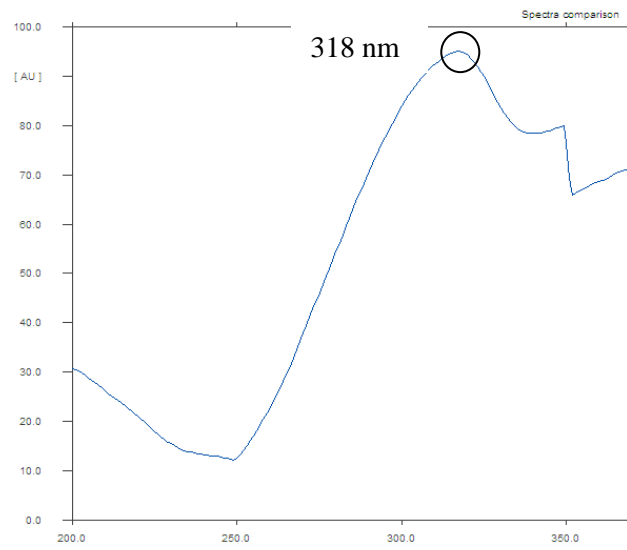


Sampel C pada fase gerak klorofom : metanol

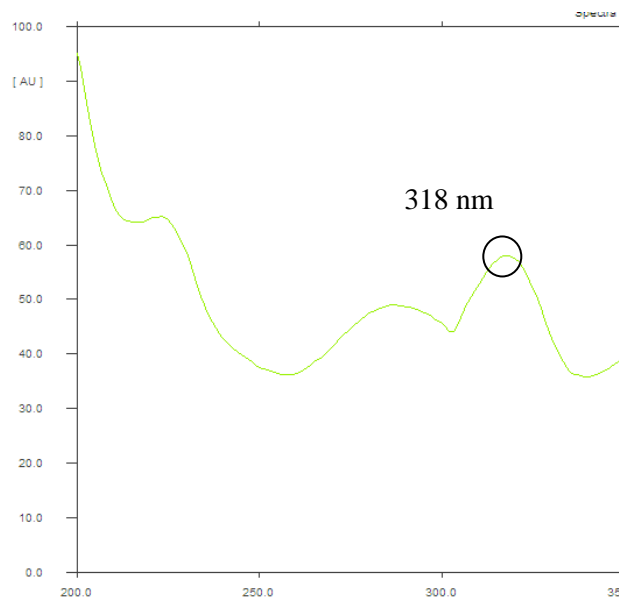
d. Sampel D



Sampel D pada fase gerak toluen:asam asetat glasial

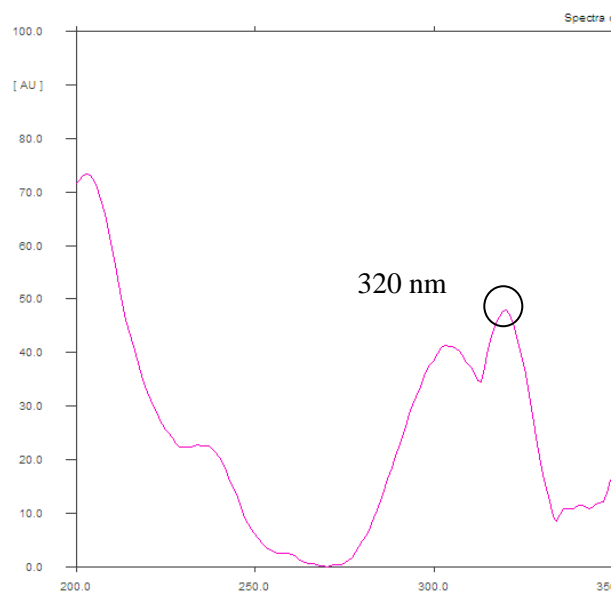


Sampel D pada fase gerak n-heksan:aseton

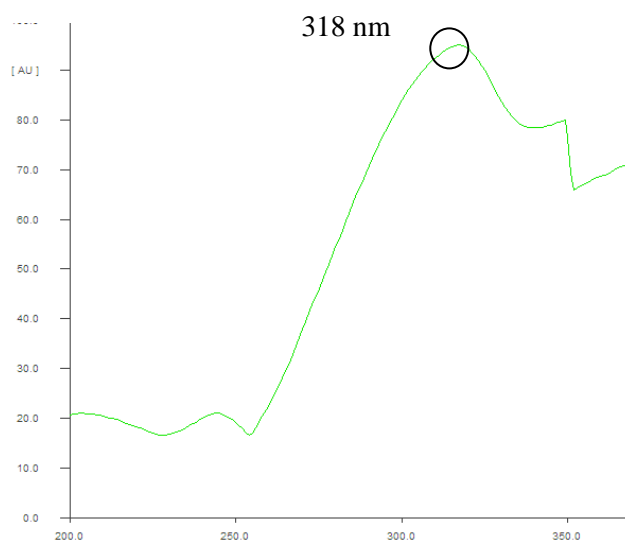


Sampel D pada fase gerak klorofom : metanol

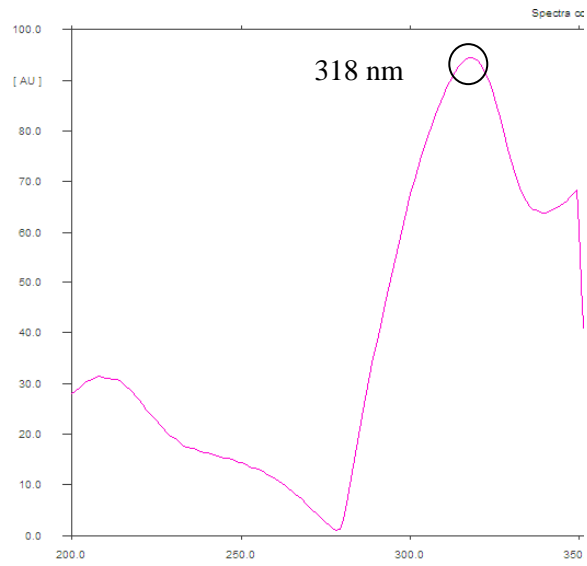
e. Sampel F



Sampel F pada fase gerak toluen:asam asetat glasial

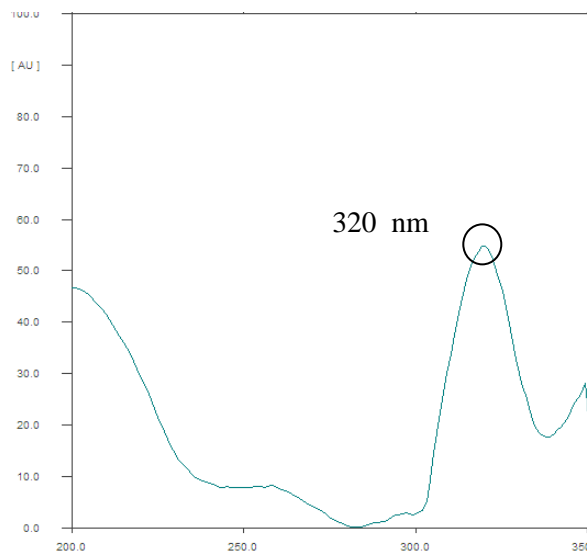


Sampel F pada fase gerak n-heksan:aseton

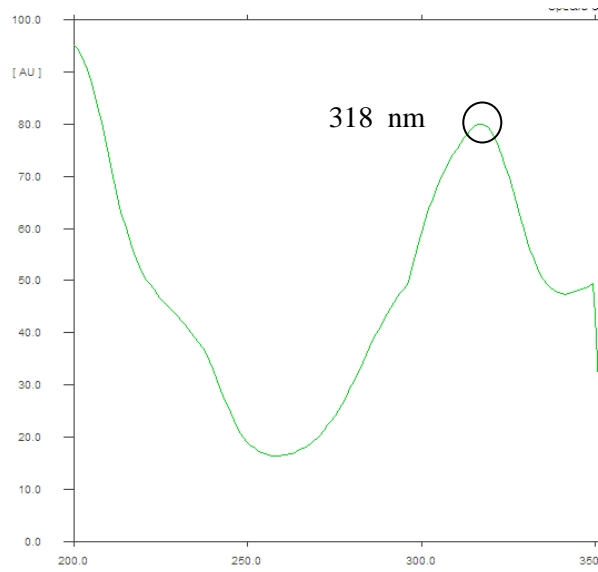


Sampel F pada fase gerak klorofom : metanol

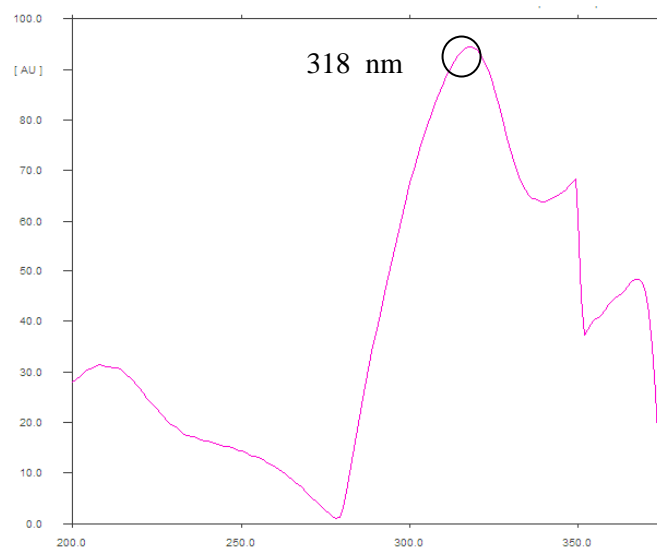
f. Sampel G



Sampel G pada fase gerak toluen:asam asetat glasial

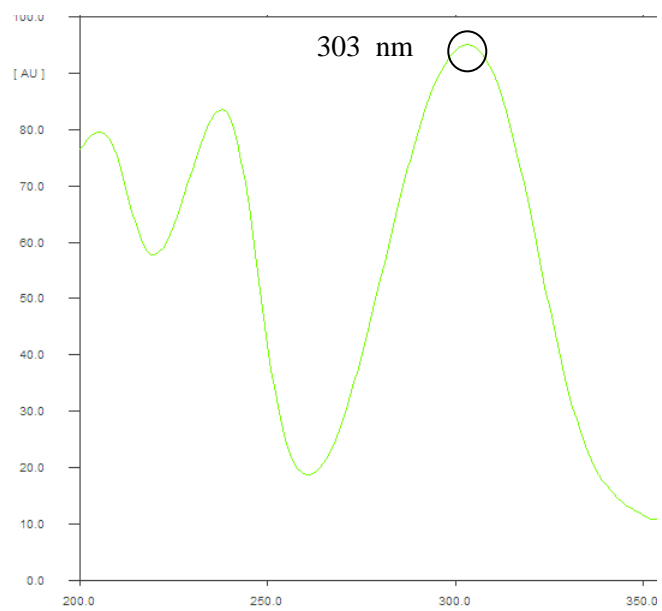


Sampel G pada fase gerak n-heksan:aseton

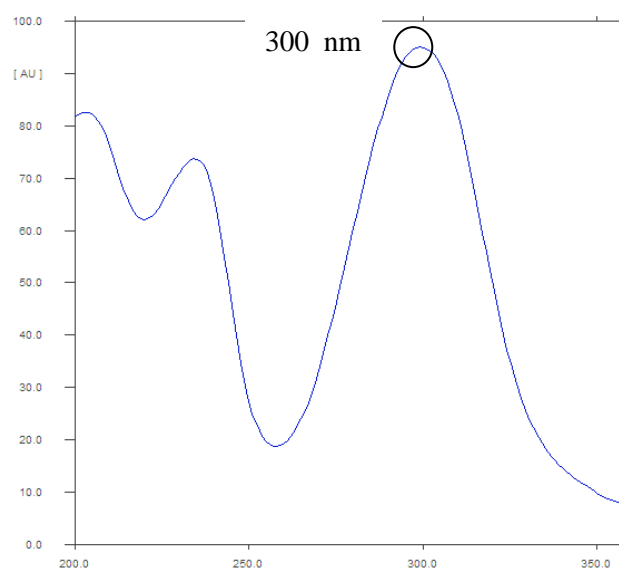


Sampel G pada fase gerak klorofom : metanol

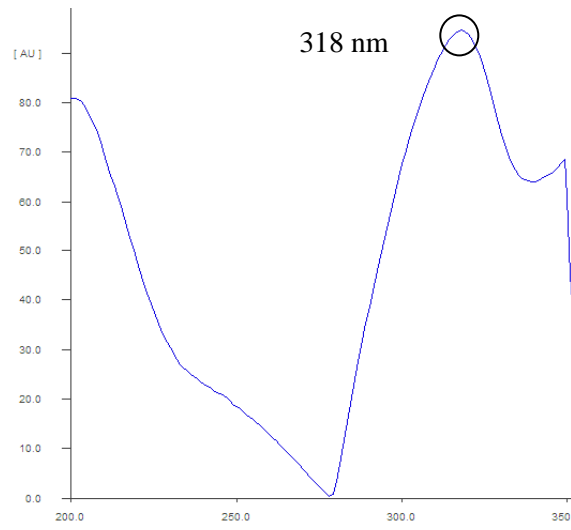
g. Sampel H



Sampel H pada fase gerak toluen:asam asetat glasial

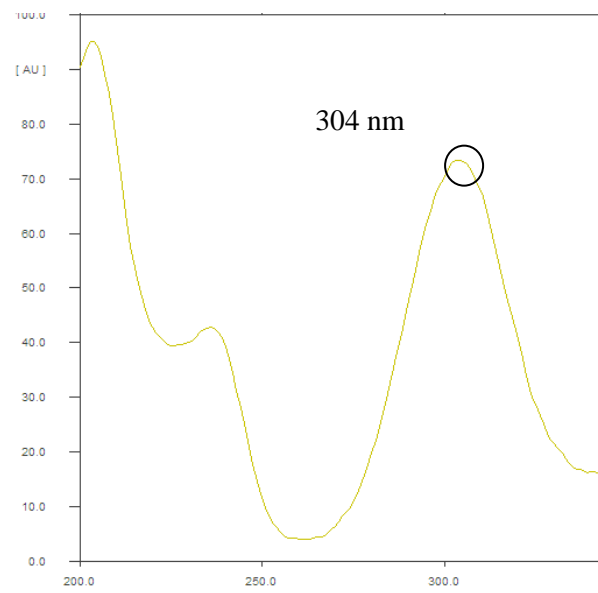


Sampel H pada fase gerak n-heksan:aseton

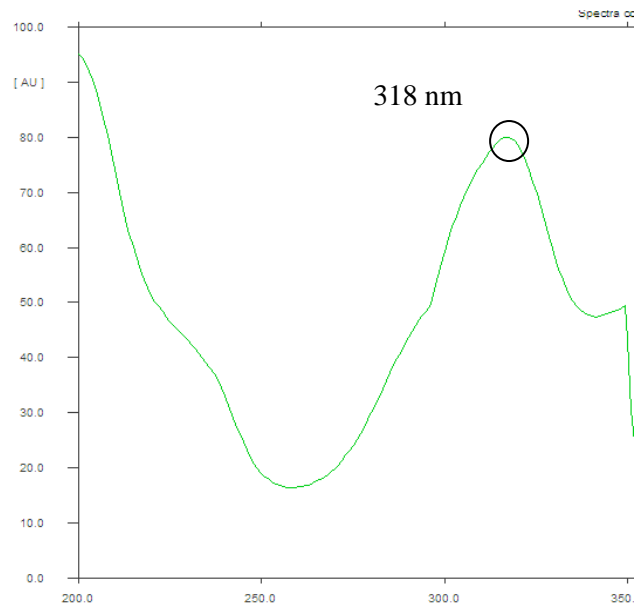


Sampel H pada fase gerak klorofom : metanol

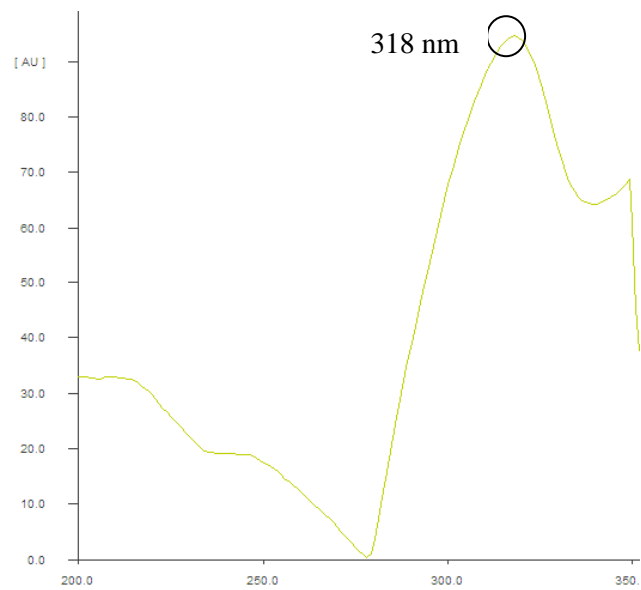
h. Sampel I



Sampel I pada fase gerak toluen:asam asetat glasial

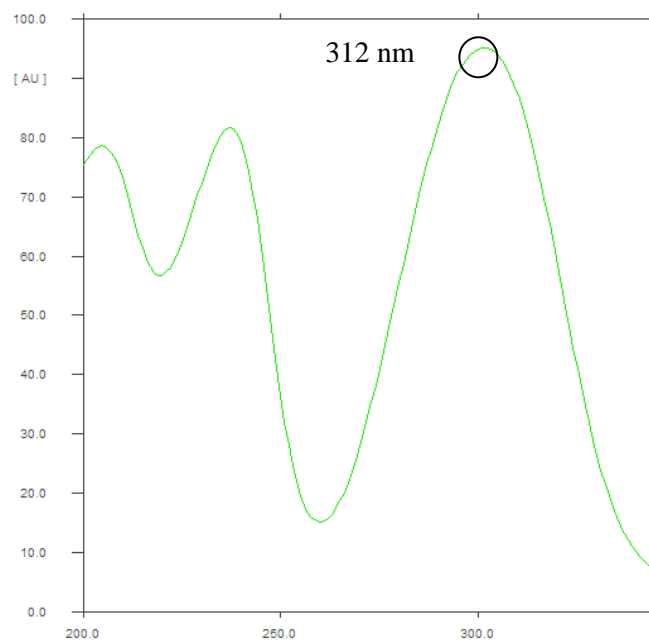


Sampel I pada fase gerak n-heksan:aseton

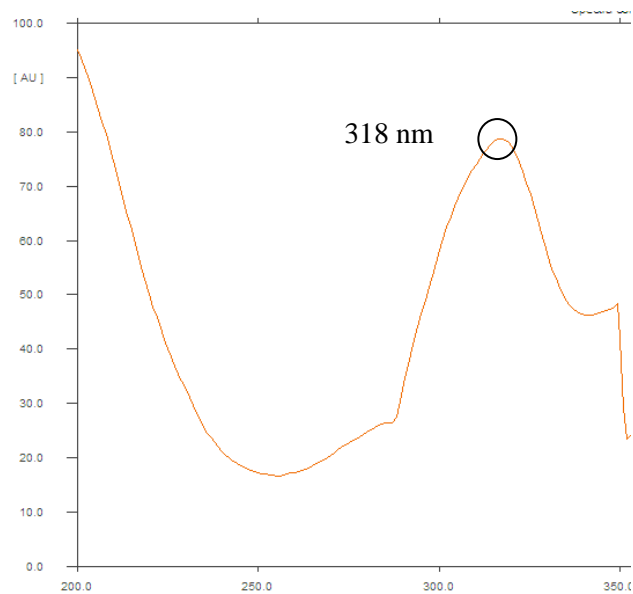


Sampel I pada fase gerak klorofom : metanol

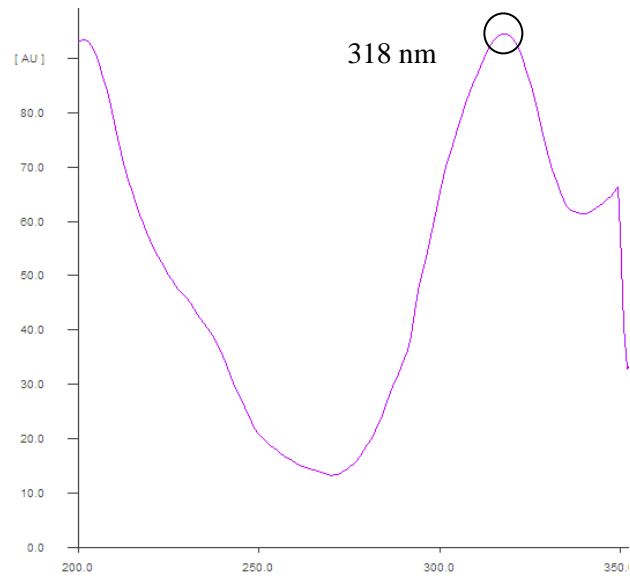
i. Sampel J



Sampel J pada fase gerak toluen:asam asetat glasial

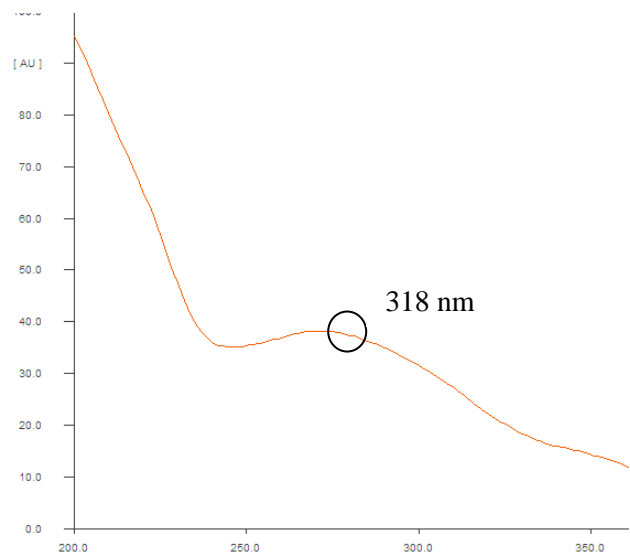


Sampel J pada fase gerak n-heksan:aseton

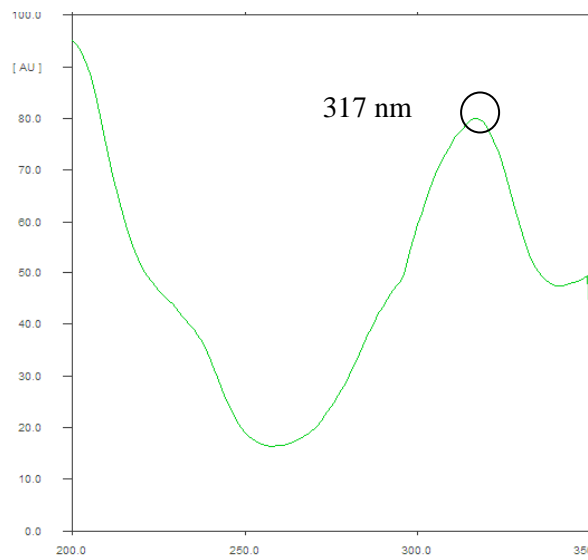


Sampel J pada fase gerak klorofom : metanol

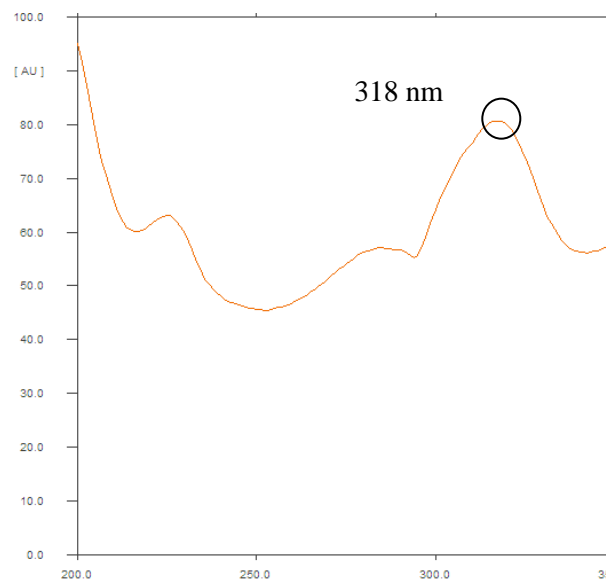
j. Sampel L



Sampel L pada fase gerak toluen:asam asetat glasial



Sampel L pada fase gerak n-heksan:aseton



Sampel L pada fase gerak klorofom : metanol

Lampiran 4. Dokumentasi



12 Krim yang Diuji



Pembuatan Larutan Uji



Densitometer yang Digunakan dalam Penelitian